

Figure 1: Logarithm of limiting activity coefficient at atmospheric pressure of furan in *n*-pentane ( $\blacklozenge$ ), *n*-hexane ( $\blacktriangle$ ), *n*-nonane ( $\bullet$ ), ethanol ( $\lozenge$ ), 1-propanol ( $\Delta$ ) and 1-hexanol ( $\circ$ ) as a function of inverse temperature.

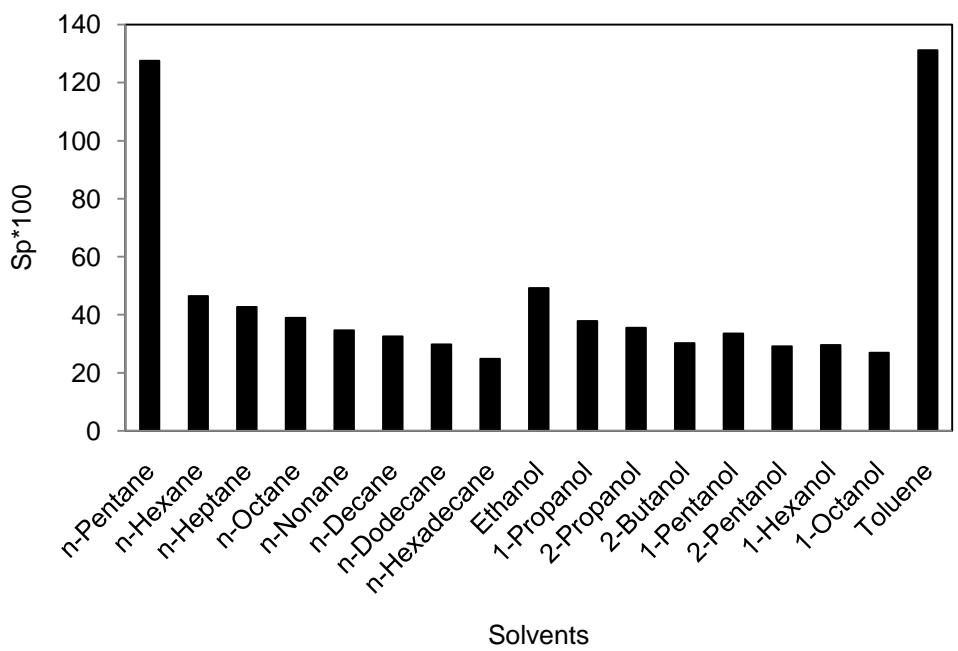


Figure 2: Comparison of solvating power values at 298 K

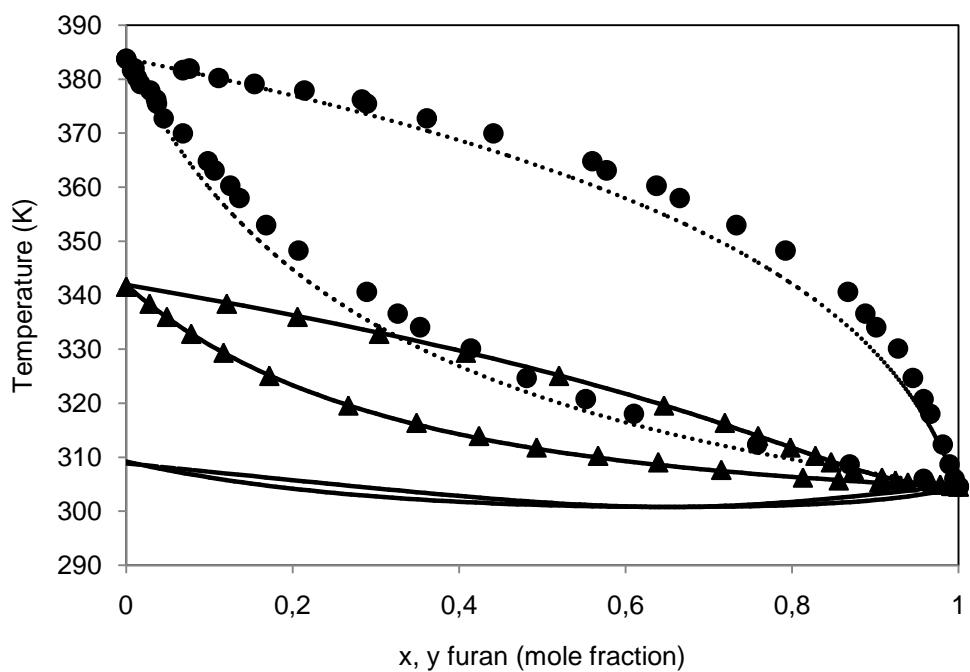


Figure 3: Vapour liquid equilibrium at atmospheric pressure for furan + *n*-pentane (dashed line), furan + *n*-hexane ( $\blacktriangle$  and solid line) and toluene + furan ( $\bullet$  and dotted line). Lines: original UNIFAC prediction, Symbols: Nala, M., et al. [5].