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# Silcretes: insights into the occurrences and formation of materials sourced for stone tool making

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# 2 Abstract

3	Silcretes are clearly observed and abundant as components of paleolandscapes on
4	several continents. Mechanisms for the formation of several varieties of silcrete, with specific
5	relationships to paleolandscapes, are described. Each type of silcrete displays particular
6	morphological features in its profile in the paleo-regolith, and these features provide pointers
7	to its origin via mechanisms of absolute or relative accumulation of silica in specific
8	environments relating to groundwater or soil-water hydrology. The characters of silcrete
9	varieties that may have triggered the interest of prehistoric peoples to exploit them for
10	manufacturing stone tools, and which control knappability, include granulometry and the
11	specific nature of silica cements. The successions of silica precipitation and recrystallisation
12	events are clearly evident as a complex of micromorphological features that provide clues to
13	the hydrological environment and its geochemistry at the time or times of silicification.
14	Examples are given of the distribution of different silcrete facies, which could have had
15	differing values for exploitation for stone tool production, in modern-day landscapes in
16	France and Australia.
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20 21	Highlights
22	Silcretes occur widely in paleolandscapes in several continents.
23	What are they and how did they form?
24 25	What made silcretes suitable for prehistoric people to form stone tools?
25 26	
20 27	
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29	
30	Keywords
31	
32	silicification, silcrete, petrography, mineralogy, landscapes, prehistoric tools.
33 24	
34 35	
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#### 36 **1** Introduction

37 There is a wealth of literature on silcretes in both the geological sciences and the 38 archaeological sciences. In geology it ranges from initial compilations of knowledge like that 39 edited by Langford-Smith (1978) to the comprehensive overview of Nash and Ullyot (2007). 40 In archaeology in Australia the studies described by Holdaway and Fanning (2014) also 41 provide a review of current knowledge and a foundation for future research. The science 42 pertaining to silcretes has actually advanced in parallel in both scientific fields with only 43 limited cross-reference, except perhaps for the work of Webb and his colleagues (Webb and 44 Domanski, 2008). In geology the early work focussed on field observations and relationships, 45 and progressed to petrographic and mineralogical studies, from which hypotheses were 46 generated about origin and palaeoenvironmental conditions. From our geological perspective, 47 some studies of stone tools in archaeological science had a 'primitive' view of silcrete, such 48 as that expressed in Mulvaney & Kamminga (1999, p 213), but others developed a 49 sophisticated understanding of the rock mechanical properties of silcrete and other siliceous 50 materials (Domanski et al., 1994), and some also used petrographic/micromorphological 51 fabrics and textures (e.g. Summerfield, 1983) to characterise silcrete artefacts and proffer 52 ideas about their provenance. But the connection with step-by-step advances in our 53 understanding of the complex of processes and environmental conditions that have led to the 54 formation of silcretes has not been maintained and the aim of this paper is to at least partly 55 address this situation.

In the first instance, we should be clear about terminology. There are several
definitions of the term 'silcrete' but we prefer something along the lines of that proposed by
Eggleton (2001), namely;

Strongly silicified, indurated regolith, generally of low permeability, commonly
having a conchoidal fracture with a vitreous lustre. Represents the complete or nearcomplete silicification of regolith by the transformation of precursor silica or silicates
and/or the infilling of available voids, including fractures. On a macroscopic scale,
some silcretes are dense and massive, but others may be nodular, columnar, blocky,
or cellular with boxwork structure. On a microscopic scale, the fabric, mineralogy

65 and composition of silcretes may reflect those of the parent material, but also indicate the changes experienced by, as well as the general environments of, silicification. 66 The key component of the definition, from our perspective, is the phrase '*strongly* 67 68 indurated, silicified regolith'. Thus, silica-rich secondary materials including flint, chert, 69 agate, chalcedony, and precious and common opal are not silcretes, though there may be a 70 spatial and temporal association with some forms of silcrete. Sandstone, quartzite and some 71 forms of porcellanite and jasper (as defined, for example, in Gary et al., 1973) that have 72 formed specifically as a result of secondary silicification of the regolith are silcretes, but those 73 resulting from diagenetic alteration or metamorphism are not. Red-brown hardpans (Chartres, 74 1985; Wright, 1983) are not silcretes because they are generally not 'strongly indurated' by 75 silica.

It is also important to note that the term '*silcrete*' is an English word and concept that has no equivalent in the geosciences literature in Latin languages where such material is generally called '*sandstone*', '*silicified limestone*', '*silicified claystone*', and so on. In France, the word '*silcrete*' was used for the first time by Parron *et al.* (1976). As well, '*flint tool*' is often used as a generic term in the French literature and could have been formed from silicified limestone or pedogenic silcrete.

82 The environments of formation and ages of silcretes have been the subject of many 83 investigations over many years, but definitive answers still elude us. Recent work in 84 Australia on the formation and age of precious opal in Great Australian Basin landscapes (Dutkiewicz et al., 2015; Rey, 2013; see also critique by Dickson, 2014 and response by Rey, 85 86 2014) did not unravel the genetic and temporal relationship between the widespread formation 87 of opal in voids and fractures in the regolith (within and outside the margins of the Great 88 Australian Basin) and the complex suite of silcrete facies in these regions. Some work on 89 O/H isotope composition and trace element chemistry of opal and the siliceous matrix in 90 silcretes (Alexandre et al., 2004; Dutkiewicz et al., 2015; Harwood et al., 2013; Webb and 91 Golding, 1998), may provide a future basis for interpreting conditions of formation, though 92 not age.

In this paper we summarise our current thinking on the characteristics, landscape
associations, origin and age of silcretes. We also attempt to link petrological and

95 mineralogical features (which are of specific importance to geologists in generating 96 hypotheses to explain the origin of silcretes) to the potential value of silcretes to prehistoric 97 peoples for the production of stone tools,. However, we recognise that there are substantial 98 fields of knowledge in geoarchaeology and lithic technology that rightly lay claim to the 99 development of an understanding of the associations between prehistoric peoples and the 100 source materials for stone tools.

101 Numerous siliceous materials used for tool making (flint, chert, quartzite, quartz) are 102 bound to geological formations which were buried, then exhumed and exposed fortuitously at 103 the surface in landscapes where they could be accessed, tested and exploited if suitable. On 104 the other hand, silcretes are directly linked to the landscapes in which prehistoric people lived. 105 Those silcretes that frame and armour landsurfaces as duricrusts and lags were highly visible 106 and available for testing and exploitation over wide areas of continents.

107 The most widely known silcrete duricrusts occur now in the desertic landscapes of 108 inland Australia, South Africa, Saharan Africa and the northeast of Brazil. They armour 109 plateaux and mesas dissected by erosion, thus maintaining steep scarps or 'breakaways' cut in 110 vari-coloured soft substrate formations that confer a spectacular aspect to the lands that 111 fascinated prehistoric people and still fascinate tourists. Silcretes are also important regolith 112 features and frame a number of landscapes in temperate regions (for example the 113 superimposed plateaux in the Paris Basin), but are less spectacular because they tend to be 114 covered and hidden by vegetation. During the glacial periods in the Late Cenozoic these 115 northern countries were somewhat bare of vegetation, with contrasting relief.

116 Although there are several forms of silcrete that can be found and recognised on many 117 continents, only some appear to have been suitable for stone tool making. The sites that have 118 generated starting materials for appropriate tools can be readily recognised in the field from 119 the litter of flakes and chips discarded during tool production. Mineralogical and 120 micromorphological properties can explain physical appearance, hardness, lustre and potential 121 durability of silcrete tools and, in some cases, in some regions, point to their provenance. 122 However, archaeologists may not be particularly focussed on sources of silcrete (or other 123 forms of siliceous rocks) that have been opportunistically exploited for the production of

stone tools, but rather seek to recognise specific source materials that have been valued from a technological, aesthetic or symbolic viewpoint and which potentially have been transported long distances. Identifying provenance from this perspective permits them to potentially trace out migration patterns, exchanges, and possibly social behaviour.

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# 2 Behaviour of silica in regolith environments

In order to explain the internal structures of silcretes, as well as their distribution and location in specific landscape settings, it is important to understand some concepts of silica behaviour in surficial environments.

132 **2.1** Silica solubility

The solubility of amorphous silica in water under earth-surface conditions is about 120 133 mg/L SiO<sub>2</sub> while the solubility of all other silica phases is significantly lower than this (Fig. 134 135 1). The most soluble phases in a geochemical system control the precipitation of the less 136 soluble ones and, in this way, amorphous silica in equilibrium with a solution is able to 137 sustain the crystallisation of every other crystalline silica phase. In turn, cristobalite, 138 tridymite, chalcedony and opal in equilibrium with a solution are able to sustain the 139 crystallisation of quartz (Fig. 1; Garcia-Hernandez, 1981; Gislason et al., 1993; Iler, 1979; 140 Siffert, 1967).

Detailed mineralogical and petrographic studies of silcretes show a consistent and
specific sequence of transformation/recrystallisation from more to less soluble phases,
namely:

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144 opal \rightarrow microcrystalline quartz \rightarrow euhedral quartz
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These transformations are the mainspring of the evolution of silcrete profiles and
especially account for the development of complex nodular structures in some forms. The
succession of recrystallization is irreversible and ultimately favours the formation of quartz
(Thiry and Millot, 1987).

150 It should be noted that most natural groundwaters around the world have a silica 151 content between 12 -18 ppm SiO<sub>2</sub> (Garrels and Christ, 1965, Swanberg and Morgan, 1978; White et al., 1963) that is roughly in equilibrium with clay minerals. This means that most 152 153 groundwaters are oversaturated with respect to quartz (solubility 4 - 7 mg/L) and are thus 154 potentially able to precipitate quartz when favourable physico-chemical conditions occur. As 155 pointed out by Thiry et al. (2014), solutions with a high silica content favour the formation of 156 crystal nuclei and crystal defects that restrict the growth of crystals and consequently the 157 precipitation of amorphous or low crystallinity silica varieties. However, in dilute solutions, 158 the number of nuclei and impurities remains limited. As a consequence, pure silica crystallites 159 generally form and large quartz crystals can develop or, under less favourable conditions, 160 microcrystalline or fibrous silica varieties are formed.

161 If silica is present in solution it has to be precipitated in order to generate silicification 162 and ultimately produce a silcrete. Several tracks are possible. It has long been thought that 163 increasing the concentration of the solution by evaporation was the principal triggering factor 164 for silica precipitation in relation to silcrete formation (Auzel and Cailleux, 1949; Millot 1970; Smale, 1973; Storz, 1928). This hypothesis resulted mainly from studies of the 165 166 widespread silcrete duricrusts in current desertic environments, for example in Australia, 167 South Africa and North Africa. It has now been established that these silcretes do not relate 168 to present day landscapes and climates and are in fact much older. Moreover, it is unlikely 169 that evaporation could be effective in developing meter-thick silicified horizons because it 170 would be much reduced, or cease, as soon as silica cementation commenced. Nevertheless, 171 evaporative processes may assist in concentrating silica in solution in near-surface 172 environments.

In contrast to evaporation, silica solubility decreases exponentially with decreasing temperature. Between 12.5 and 0°C, the solubility of quartz decreases to half its initial value (Fig. 2). This appears to be a very effective factor in precipitating silica from solution: it had never been envisaged as a mechanism for precipitating silica in the supergene realm until proposed by Thiry *et al.* (2013, 2015b) but was commonly advocated in hydrothermal environments. 179 Increasing the concentrations of salts in solution is another efficient way to decrease 180 silica solubility: the solubility of amorphous silica decreases with increasing salt 181 concentration to saturation (Fig. 3), up to 96% in the case of CaCl<sub>2</sub> and up to 30% in the case 182 of a NaCl saturated solution (Marshall, 1980). The mixing of a silica solution with chlorite-183 and/or sulphate-rich brine is certainly a very effective mechanism for inducing the 184 precipitation of silica. The often advocated relation between silicification and warm and dry 185 climates most probably results from the negative relationship between increasing 186 concentrations of evaporite brines and silica solubility.

# 187 2.2 Key geochemical processes in regolith solutions

As silica in regolith solutions mainly originates from the weathering or alteration of alumino-silicates (such as feldspars and clay minerals) in rocks and sediments, the relative solubilities of Si and Al must be taken into account. Si and Al have different domains of stability (Fig. 4).

(1) In acidic environments, Al and Si have contradictory behaviours. Al is more
soluble than Si, and so alumino-silicates (clays and feldspars) are destroyed and leached of Al
whereas Si in the silicate frameworks, and silica minerals, remain essentially *in situ*, most
often in the form of opal;

(2) At neutral pH, Al and Si behave similarly. Al and and Si both have a lowsolubility. Both silicates and silica are relatively stable in this domain;

(3) In alkaline environments Al and Si are both highly soluble. Silicates and silica
may be altered, but such environments are rare and restricted in continental landscapes (for
example alkaline lakes in the east African graben).

Highly acidic conditions appear to be especially favourable for the degradation of silicates and clay minerals via dissolution and leaching of Al and other cations to concentrate silica. However, such acidic environments are not common in regolith environments and require specific conditions to develop. Sulfide oxidation is one possibility for the development of acid conditions. It is a
consequence of weathering of sulfides present in host rocks, resulting in high concentrations
of sulfur in groundwaters, abundant sulfate minerals and an oxidizing environment.
Nevertheless, pyrite oxidation will only develop widespread acidity if water flow through
these formations is limited. Ultimately, the acidity depends on the degree of oxidation
reached:

211 (1) 
$$\operatorname{FeS}_2 + 7/2 \operatorname{O}_2 + \operatorname{H}_2 \operatorname{O} \rightarrow \operatorname{Fe}^{++} + 2 \operatorname{SO}_4^{--} + 2 \operatorname{H}^+$$

212 (2) 
$$2 \operatorname{FeS}_2 + \frac{15}{2} \operatorname{O}_2 + 7 \operatorname{H}_2 \operatorname{O} \rightarrow 2 (\operatorname{Fe}(\operatorname{OH})_3 + 4 \operatorname{SO}_4^{--} + 8 \operatorname{H}^+)$$

213 Ferrolysis is another geochemical process that develops highly acidic environments. It 214 relies on the oxidation of Fe<sup>++</sup>-bearing solutions by incoming  $O_2$ , leading to the liberation of 215  $H^+$  protons (Fig. 5).

216 (3) 
$$\operatorname{Fe}^{++} \to \operatorname{Fe}^{+++} + 2e^{-}$$

217 (4) 
$$2Fe^{+++} + 4H_2O \rightarrow 2FeO(OH) + 6H^+$$

It has been widely advocated in several soil environments in which clay minerals are
degraded by leaching of Al from the silicate framework (Brinkman, 1970). In Western
Australia, pyrite oxidation and ferrolysis have been put forward to explain the regional acid
saline groundwater systems and hundreds of ephemeral saline lakes characterized by complex

acid brines with a pH as low as 1.4, together with an abundance of alunite, and the large-scale

223 mobilisation of silica and accompanying silcrete formation (Benison and Bowen, 2015;

Dickson and Giblin, 2009; Mann, 1983; McArthur *et al.*, 1991).

The effectiveness of acidolysis of alumino-silicates (feldspars and clay minerals) with regard
to silicification can be appreciated from the mass balance equations:

(5) 
$$2 (\text{KAlSi}_{3}\text{O}_{8}) + 2 \text{H}_{2}\text{O} + 2 \text{H}^{+} \rightarrow \frac{\text{Si}_{2}\text{Al}_{2}\text{O}_{5}(\text{OH})_{4} + 4 \text{Si}_{2}\text{O}_{2} + 2 \text{K}^{+}}{\text{kaolinite 46 g}}$$
(6) 
$$\frac{\text{Si}_{2}\text{Al}_{2}\text{O}_{5}(\text{OH})_{4} + 6 \text{H}^{+} \rightarrow 2\text{Al}^{+++} + 2 \text{Si}_{2}\text{O}_{2} + 5 \text{H}_{2}\text{O}}{\text{kaolinite 46g}}$$
(6) 
$$\frac{\text{Si}_{2}\text{Al}_{2}\text{O}_{5}(\text{OH})_{4} + 6 \text{H}^{+} \rightarrow 2\text{Al}^{+++} + 2 \text{Si}_{2}\text{O}_{2} + 5 \text{H}_{2}\text{O}}{\text{silica 30 g}}$$

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The complete acidolysis of orthoclase releases 73% by weight of silica: this could remain *in situ* to contribute to the formation of a silcrete (*relative* accumulation of silica) or be released in solution to infiltrate, and initiate silicification of, specific zones or horizons in the regolith
(*absolute* accumulation of silica).

#### 232 **2.3** Silicification in the regolith

In general, silcretes have formed as a result of either *absolute* or *relative* accumulationof silica.

Relative accumulation of silica occurs primarily as the result of the dissolution and 235 236 leaching of alumino-silicate minerals, specifically silicates such as feldspars and clays (Rayot 237 et al., 1992; Thiry et al., 2006). The process requires an acidic environment that is aggressive 238 towards these minerals and generates the loss of non-Si elements, resulting in an in situ 239 concentration of silica. There appears to be no significant transport of Si within the system 240 but only limited local reorganization and adjustment. However, leaching of cations other than 241 silica generates a significant loss of material leading to the disruption and collapse of primary 242 fabrics and structures.

243 Absolute accumulation occurs through importation of silica to the system. The silica 244 may replace pre-existing minerals or simply fill voids, fractures and intergranular pores. The 245 silica has to originate from somewhere else and has to be brought into the horizon that 246 undergoes silicification. With regard to the relatively low solubility of silica in surficial 247 waters and the relatively large volume of silica needed to infill pore spaces to effect 248 silicification, this kind of silica accumulation requires large volumes of water to flow through 249 the horizon. This is only conceivable in groundwater outflow zones (springs, seeps) and 250 implies landscape incision in order to intersect groundwater tables. Primary geological 251 structures in the host horizon are characteristically preserved.

252

3

#### Varieties of silcrete

253 Silcretes take a wide variety of forms and detailed investigations of these forms from 254 the field scale to that available via scanning microscopy provide clues to their origin. Aspects 255 such as field distribution, geomorphological patterns and relationships to current and former 256 landscapes, petrology and mineralogical-geochemical characteristics have all to be 257 considered. We group silcretes into two classes: groundwater silcretes that form mainly as a 258 result of the absolute silica accumulation of silica in regolith environments; and pedogenic 259 silcretes that reflect the relative accumulation of silica, as well as a complex of dissolution, 260 eluviation and illuviation processes typical of soil environments. Nash and Ullyott (2007) 261 have suggested a sub-classification of 'non-pedogenic' silcretes into 'drainage-line' and 262 '*pan/lacustrine*' types', but these forms can be encompassed satisfactorily within groundwater 263 silcretes from our perspective. Of particular interest in the landscape context is that volume is 264 mostly conserved during the formation of groundwater silcretes, whereas volume is 265 substantially lost during the formation of pedogenic silcretes, some of which can be metres 266 thick in profile.

#### 267 **3.1** Groundwater silcretes

The main characteristics of groundwater silcretes are their disposition as a series of superposed silicified lenses in the regolith and the preservation of primary host-rock structures (stratification, bioturbation, fossils). There are two main varieties: those that occur in sands and develop into massive quartzitic facies by infilling and cementation of pore spaces between detrital quartz grains; and those that are created by epigenetic replacement of limestones, claystones and even gypsum horizons, and therefore have more irregular and discontinuous shapes.

275

# 3.1.1 Quartzitic types

These are typically formed in sands by cementation of the detrital quartz grains via syntaxial quartz overgrowths, or other forms of silica, precipitated in pore spaces. The initial condition is always a porous sediment. Quartzitic silcretes generally form well-confined flatlying pans within unconsolidated or poorly cemented sands. Those in the Fontainebleau Sand have been studied in detail and are good examples.

In the Paris Basin, flat-lying quartzitic silcrete lenses (1-8m thick) occur at different levels within the Fontainebleau Sand formation (Thiry *et al.*, 1988; Thiry *et al.*, 2013; Thiry *et al.*, 2015a). The silcrete lenses typically crop out on the edges of plateaux or in valley margins but pinch out rapidly under the limestone-capped plateaux, as shown by drill holes and observations in sand quarries (Fig. 6). This strong link between silcrete distribution and present-day geomorphology suggest that these silcretes developed relatively recently in and near outcrop zones of the Fontainebleau Sand. Sand calcite contained in the silcretes date the silicification to Pleistocene glacial periods (Thiry *et. al.*, 2013).

Geomorphological and geological constraints suggest that silicification occurred in
groundwater outflow zones (springs, seeps) during landscape incision (Thiry *et al.*, 1988;
Fig. 7). The key process sequence is:

precipitation of silica in horizons of the Fontainebleau Sand in a zone of groundwater
outflow; and

• resumption of erosion leading to lowering of the groundwater table and the formation of
 a new silcrete pan via groundwater outflow during a subsequent period of landscape
 stability;

The superposition of separate silcrete pans rather than the formation of a continuous zone ofsilicification reflects periodicity in valley down-cutting and landscape stability.

The silica necessary for cementation of the Fontainebleau Sand to form quartzitic silcrete has come from groundwater contained within the formation. Its silica content ranges from 10 - 15 ppm, which is about three times the saturation value of quartz (4 - 6 ppm) at surface temperatures and pressures.

303 The correlation of silica cementation with glacial periods, as determined by age dating 304 of contained calcite crystals, suggests that silica precipitation occurred by cooling of 305 groundwaters along discharge pathways towards outflow zones in frozen soils in the valleys. 306 Hydraulic and geochemical mass balance calculations, based on the silica content of water in 307 an average current-day spring, indicate that cementation could potentially take place over 308 about 2 000 years. This is a short geological time span and totally changes previous 309 interpretations of the time-scale for the formation of some silcretes. 310 Similar quartzitic silcretes appear to have formed under cold environments during the

Pleistocene in North America (Ludwig and Paces, 2002; McCoy, 2011), and such an origin

312 may apply to other occurrences, like those within white sandstone formations in Germany

313 (Götze and Walther, 1995) and in the Czech Republic (Mikulas, 2002), Greece (Skarpelis, 314 2006), southern France (Parron et al., 1976) and Spain (Bustillo and Bustillo, 2000; Parcerisa 315 et al., 2001), for which the origins have not yet been considered in detail. Quartzitic silcrete 316 pans may also develop in very different environments, including those that have not 317 apparently experienced cold climates. For example, in Australia (Thiry and Milnes, 1991) 318 and southern Africa (Summerfield, 1983; Nash et al., 1998), quartzitic silcretes may have 319 formed in groundwater environments in dry paleoclimates where silica precipitation could be 320 attributed to a mixing of fresh and saline waters. Nevertheless, all examples relate to 321 groundwater outflows in incised landscapes and for which silica has been imported in order to 322 provision precipitation and cementation.

323

#### 3.1.2 Silicified limestone/calcrete types

Many lacustrine limestone formations contain irregular silicified masses and have long been described from dry regions such as the Kalahari Desert, North Africa and Australia, as well as in more humid countries like France, Spain and the USA (Alimen and Deicha, 1959; Banks, 1970; Daley, 1989; Kaiser, 1928; Nash *et al.*, 2004; Thiry and Ben Brahim, 1997). The silica was generally thought to have precipitated during dry periods at the time of limestone deposition in confined environments. However, petrographic observations and geochemical data have generated new ideas about the formation of these silcretes.

The extensive Tertiary lacustrine limestones of the Paris Basin form superimposed plateaux that constitute the dominant geomorphologic features. Typical examples are the Brie and Beauce plateaux. Almost all of these limestones contain zones of silicification (Cayeux, 1929). Their size varies from millimetre-sized dots to siliceous bodies several tens of meters long. The silicified facies have an irregular distribution (Fig. 8): locally, they can form over 10% of a quarry cliff-face, but are absent a short distance away. On the regional scale they average about 5% of the volume of the lacustrine limestone in the southern Paris Basin.

Two types of silicification coexist: deposits of quartz and chalcedony in voids; and epigenetic replacement of the limestone matrix by microcrystalline quartz, with preservation of the primary sedimentary structures. Thin sections show a systematic link between silicification and zones of high porosity, either preserved or partly infilled by silica deposits(Thiry and Ribet, 1999).

343 Because the silicified limestones are pure, without any sandy or clayey layers, silica 344 had to be imported from elsewhere. And as the limestones have never been deeply buried, the 345 importation of silica can only have been via substantial near-surface groundwater flows, 346 which could explain the relationship between porosity and the intensity and extent of the 347 silicification. This is a very different mechanism of silica accumulation from that represented 348 by flints and cherts. In this case, silicification occurs during burial diagenesis whereby there 349 is epigenetic replacement of limestone by silica originating in microfossils in specific 350 sedimentary layers, without the formation of any voids.

351 Substantial groundwater outflows are only possible after uplift and incision of the 352 limestone formation. In the Paris Basin, such conditions were only fulfilled during the 353 Pliocene and the Quaternary periods and, therefore, the silicification had to occur relatively quickly (Thiry and Ribet, 1999). Similar groundwater silicification of limestone has already 354 355 been described in present day hydrologic systems in inland Australia (Arakel et al., 1989; 356 Benbow, 1993), southern Africa (Nash et al., 2004) and Spain (Armenteros et al., 1995). As 357 for quartzitic silcretes, silica precipitation was probably triggered by different geochemical 358 mechanisms (water cooling or water mixing, or others), under very different climatic and 359 environmental conditions. However, the resulting petrographic fabrics are similar: dissolution 360 of calcite making space for precipitation of silica, which may lead to very contrasted fabrics.

Weathering of Paris Basin silicified lacustrine limestone during Pliocene-Quaternary times produced cavernous silicifications called "meulières" which delineate plateau surfaces in the southern Paris Basin (Beauce Plateaux; Thiry, 1999). The meulières show a large variety of facies from massive to highly porous 'cellular' facies and can be weathered and desilicified into friable blocks, even breaking down into a rough sand-like material.

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#### 3.1.3 Porcellanite/jasper types

Fine grained silcretes, obviously resulting from silicification of primary clay-rich materials, form specific facies, generally of glassy appearance, and often resembling flint or chert except for their colour which is generally due to iron-oxide inclusions. Such materials have been described in several geological contexts (Boule, 1888; Millot *et al.*, 1959, Valleron,
1981).

372 Such silicifications are widespread in inland Australia and are commonly associated 373 with bleached and weathered profiles in rocks and sediments ranging in age from Precambrian 374 to Quaternary (Simon-Coincon et al., 1996). The profiles are up to 60m deep. Their main 375 characteristics are white or pastel colour and low bulk density, together with the occurrence of 376 alunite, iron oxide mottling, extensive bioturbation by termite burrows that penetrate to great 377 depth, and ubiquitous gypsum that has crystallised in fractures and voids (Thiry et al., 1995; 378 Thiry et al., 2006). The bleached material is mainly formed of quartz, large booklet-like 379 kaolinite and opal A, with alunite and gypsum in pods and veins (Fig. 9). In some localities 380 the burrows and their fillings, as well as the enclosing saprolite, have been silicified. K/Ar 381 dating of alunite in these profiles indicates a Miocene age for the bleaching and weathering, 382 varying from about 16 - 18 Ma in the Coober Pedy opal field to about 8 - 12 Ma in the 383 Andamooka opal field (Bird et al., 1990).

384 These profiles have resulted from alteration under acidic conditions in a saline 385 groundwater environment nearly saturated in gypsum (Thiry et al., 1995). The bleaching was 386 related to the lowering of the groundwater table. Arid conditions at the time restricted water 387 flow and allowed an acidic environment to develop to leach original alumino-silicates, retain 388 Si (as amorphous silica and quartz), but not completely lose some of the most soluble 389 elements like K, Ca, SO<sub>4</sub> and Al from the system that form nodules and veins of alunite and 390 gypsum in places. Elevated concentrations of anatase (Fig. 9) could indicate that Ti released 391 from primary alumino-silicates has also been retained.

Horizontally disposed lenses or pods of porcellanite or jasper subsequently developed in the bleached saprolite (Fig. 9). Primary sedimentary structures and fabrics, as well as some fabrics relating to weathering and bleaching, in particular iron-mottling, have been retained within these silica-indurated horizons. The superposition of several silica-indurated layers relates to still-stand positions of paleo-water tables which rose significantly after the widespread bleaching, thus reflecting a return to more humid climatic conditions.

398 Low concentrations of clays in the porcellanite and jasper horizons compared with the 399 host regolith (Fig. 9) points to acidic environments that probably developed via ferrolysis and 400 generated silica via relative accumulation. However, micromorphological examination shows 401 that the silicified horizons are cemented by various silica phases. The silica cements were 402 precipitated from water inflow, either by infiltration through the upper part of the profile, or 403 by lateral flow of groundwater containing silica through the profile. The silica was probably 404 derived from dissolution of silica (opal-A, quartz) and silicates (clays) in the bleached profile. 405 The formation of these groundwater silcretes required a consistent hydraulic flow regime that 406 was probably related to down cutting of the landscape (Simon-Coinçon et al., 1996). Silica 407 precipitation points to mixing of fresh silica-charged water with sulphate-rich groundwater, 408 thus lowering silica solubility.

409 **3.2** *Pedogenic silcretes* 

Pedogenic silcretes developed near the landsurface, within soil environments, and
display typical soil structures (geopetal features) related to infiltration of downward
percolating water (such as differentiated horizons, eluviation and illuviation structures).
Details of this silcrete type come mainly from Eocene occurrences in the Paris Basin (Thiry,
1981). Specific characteristics and mechanisms of formation are also derived from
observations of similar materials in Australia (Milnes and Thiry, 1992; Simon-Coinçon *et al.*,
1996; Thiry and Milnes, 1991) and Central America (Elsass *et al.*, 2000).

In the southern part of the Paris Basin, silicified pans and lenses cap Eocene detrital kaolinitic deposits. These are very hard, tightly cemented quartzitic silcretes that break up into variably sized blocks, with puffy, tear-like shapes, coated with deposits ('cappings') of yellow-white opaline silica that mask the internal structures. Typically, the silcretes have a columnar structure with characteristic laminated cappings on the columns (Fig 10). They are about 2 m thick and display several distinct horizons with systematic micromorphological and mineralogical structures (Thiry, 1981).

A lower granular horizon consists of a sandy claystone with millimetric- to
 centimetric-sized granules of microcrystalline quartz and opal. TEM and electron diffraction

studies of the clay matrix show that hexagonal-shaped kaolinites have corrosion embaymentsand are coated and welded together by a silica gel.

428 An overlying columnar horizon has a matrix composed of detrital quartz grains with 429 irregular overgrowth apophyses which grade into titania-enriched microcrystalline quartz. 430 Ghosts of granules are easily recognizable within the matrix. Illuviation cutans of opal have 431 developed in fractures. The youngest laminae, at the top of the illuviation structures, always 432 consist of opal with a low titania content. The initial laminae, at the base of the illuviation 433 structures, commonly have a nodular structure and are formed of microcrystalline quartz with 434 a higher titania content. This sequence indicates a progressive recrystallisation of the opal. 435 Recrystallisation is accompanied by loss of silica, which induces the destruction of the 436 primary structures and relative accumulation of titania.

A more massive horizon, with complex nodular and pseudo-breccia fabric, is found at
the top of the profile. It is devoid of opal and the microcrystalline quartz matrix has partly
dissolved. Titania has accumulated in rims around the nodules. Euhedral quartz has developed
in the voids and some quartz grains show overgrowths.

441 Pedogenic silcretes occur widely in central and southern Australia and have been 442 described in detail from the opal fields (for example Fig. 9, near-surface horizon) and the 443 region around Alice Springs (Milnes and Thiry, 1992; Milnes and Twidale, 1983; Simon-444 Coincon et al., 1996; Thiry and Milnes, 1991). Their morphology is typically that described 445 by Thiry and colleagues from around the Paris Basin (Barrois, 1878; Thiry, 1981), with wide 446 variations in the appearance of the columnar facies being evident in different localities. As 447 well, pedogenic silcretes occur widely in north-western Europe, in Belgium on the Ardennes 448 (Gosselet, 1888), in Germany on the Slate Mountains massifs (Lange; 1912; Teichmüller, 449 1958) and in the London Basin (Kerr, 1955, Summerfield, 1980). They armour South African 450 and Botswanan landscapes where Summerfield (1983) described highly variable textures and 451 fabrics classifying the main petrographic characteristics as 'grain-supported fabric' (quartzitic 452 silcrete with quartz overgrowths), 'floating fabric' (silicified clayey matrix with coarse 453 detrital grains), and 'glaebular fabric' (containing illuviation structures typical of soils).

454 In terms of the origin of pedogenic silcretes, opal initially formed in the granular 455 horizon and at the base of joints in the columnar horizon where water circulation was slow 456 and stagnation had occurred. Higher in the profile, the microcrystalline matrix dissolved. 457 Well-crystallised quartz crystals formed at the top of the profile. Profiles clearly show a 458 migration of silica from top to bottom and the inheritance, in the uppermost horizons, of 459 micromorphological features which initially developed in the lowermost horizons. This 460 inheritance demonstrates that silicification progressively 'eats' into the landscape like a 461 weathering front. The greatest amount of silica comes from dissolution at the top of the 462 profile followed, from top to bottom, by a sequence of precipitation and re-dissolution events.

The close link between leached and confined environments does not imply a strict
synchronism of degradation and construction. The two systems work in an alternating fashion,
whereas periods of loss (leaching) and accumulation (precipitation) follow one another more
or less sequentially.

Initially silica originated from the degradation of clay minerals at the base of the
profile, possibly in a highly acidic environment triggered by ferrolysis. The mineral sequence,
from top to base of the profile, results from a progressive concentration of silica and other
cations in infiltrating and downward-moving solutions. The silica content of the solution
could be concentrated by 2 to 4 times through evaporation during dry periods and silica
precipitation favoured by increasing cation concentration.

473

4

# Characteristics of silcretes of relevance to tool making

474 Silcretes show numerous variations in their morphology and their spatial arrangement 475 but also vary in their composition and internal texture. A classification of silcretes based on 476 micromorphological fabrics by Summerfield (1983) has been used widely in archaeological 477 studies to identify different types of silcrete artefacts. Indeed, each silcrete type may display 478 various types of cement and porosity, with sometimes decimetric scale variation, but this does 479 not relate to their convenience or value for tool making. Particular facies, with interesting 480 mechanical properties or appearance could exist within larger masses and may have been 481 specifically sought and used.

482 Various crystallization fabrics have developed according to the nature of the host 483 formation and/or the chemistry of the silica-bearing solutions. The nature of silica cements 484 and the geometrical relations between crystals are the predominant determinants of the 485 mechanical properties of silcretes. Substantial research in this field has been undertaken in 486 Australia from an archaeological perspective (e.g. Domanski et al., 1994; Webb and 487 Domanski, 2008). These investigations have determined that tool-making depends on two 488 main characteristics: the hardness of the material to form cutting and wear-resistant tools, and 489 the capacity to produce a regularly curved conchoidal fracture that permits the detachment of 490 long and thin knappings.

491 It is not within our capability to examine the knappability of silcrete types, nor to 492 make reference to the variety of silcrete facies used by prehistoric peoples to make stone 493 tools. Much of the formal understanding of the mechanical properties of silcretes and other 494 rocks from which stone tools were made, and the influence on these properties by heat 495 treatment, which is believed to have been commonly practiced, has been comprehensively 496 detailed (e.g. Domanski et al., 1994; Domanski and Webb, 1992; Domanski and Webb, 2007; 497 Webb and Domanski, 2008). Our objective is to illustrate how some petrographic fabrics of 498 silcretes, which we use to help unravel their origin and environment of formation, may effect 499 knappability.

500 Broadly speaking, silcretes that have the finest grain size and the least porosity display 501 the best developed conchoidal, vitreous fracture surfaces. From a petrological and 502 mineralogical perspective, the finest grain size will occur in those silcretes formed 503 predominantly of opal, which can be a complex of hydrated, poorly crystallised silica phases 504 (opal-A, opal-CT) and is consequently comparatively soft (Mohs hardness ~5-6). In those 505 silcretes in which the matrix is dominantly microcrystalline quartz, of which there are several 506 types ranging from granular to fibrous varieties, the Mohs hardness is that of quartz (7) and, 507 depending on the microporosity, fracture surfaces can be glassy in appearance. As the 508 abundance of detrital framework quartz grains increases, for example in pedogenic silcretes 509 and some quartzitic groundwater silcretes, the smoothness and lustre of fracture surfaces is

controlled essentially by the bonding between framework grains and matrix silica, and can beinfluenced by the abundance of other constituents including micro-cryptocrystalline anatase.

#### 512 4.1 'Sandstone' cementation

513 In clean sands, without a clay matrix, such as the Fontainebleau Sand, precipitated 514 silica builds on the crystal structure of each detrital quartz grain, leading to the development 515 of euhedral crystal faces. On continued precipitation, crystal faces come into contact, and 516 finally form triple point junctions at the site where the pore space has closed (Fig. 11). The 517 overgrown and expanded quartz grains are not welded together, but only meshed in a compact 518 way. This provides cohesion and hardness to a quartzitic silcrete formed in this manner. 519 When broken by hammering or knapping, the fractures can cross cut quartz grains but tends to 520 follow the euhedral crystal faces of the overgrowth quartz that make contact between the 521 grains. This results in a rough fracture surface of sugary appearance. Such fractures are not 522 suitable for making thin blades or sharp cutting edges.

523 Cementation of quartz grains may also be achieved as the result of precipitation of 524 silica in pore spaces between the grains (Fig. 11), either in the form of opal (quasi-amorphous 525 silica) which may have later re-crystallized into chalcedonite, or direct precipitation of 526 chalcedonite sheaves. These silica deposits give rise to stronger cementation than quartz 527 overgrowths and this is linked to the bonding between the detrital quartz grains and the silica 528 precipitates and the complex intercrystallite connections in the pore spaces. Accordingly, the 529 fracture in these silcretes tends to be 'clean' and cuts across the detrital quartz grains. When 530 the detrital quartz grains are relatively small and the silica deposits occupy the main part of 531 the porosity, the fractures may even be conchoidal and lustrous. This is most likely when the 532 pore fillings are of microcrystalline quartz.

#### 533 4.2 Silicified limestones

534 In silicified limestones the carbonate matrix has been replaced by silica. The 535 substitution is mainly epigenetic, which means that primary limestone structures such as 536 nodules and fossils are replaced by silica and conserved. For this to occur, there can be no dissolution of the carbonate before replacement: substitution occurs step by step along a clear
'alteration front' between existing limestone and incoming silica. This replacement implies
concomitant leaching of the carbonate and precipitation of silica. Voids are necessary for
water circulation and their shape and size determines the density and 'quality' of the
silicification.

If the voids have the form of micro-karst, which develops as the silica precipitates (Fig. 12), the resulting silcrete could be very dense, as translucent as flint and with a lustrous conchoidal fracture. On the other hand, when the dissolution features are larger, or when carbonate residues remain as impurities in the silicified zones, then the resulting silcrete has a dull aspect and a rough surface and is much less convenient for tool making.

547 Quartz is generally the main silica mineral and occurs as both epigenetic replacement 548 and also euhedral overgrowths in pores. Some of the fine microcrystalline matrix may result 549 from recrystallization of primary opal. Chalcedonite and laminated opal deposits occur in 550 pores in some sites (Fig. 12). The presence of opal and poorly crystallized forms of silica 551 may be favourable for heating transformation of these silcretes which is sometimes observed 552 naturally in outcrop in Australia.

#### 553 4.3 Porcellanites and jaspers

554 These materials are of fine grain size and are formed mainly of various petrographic 555 varieties of opal. The matrix resulting from the alteration of the primary material is generally 556 formed of milky opal made opaque by impurities and microporosity. The most compact 557 samples, with a glossy break, also contain translucent concretionary opal that cements the 558 pores within the matrix (Fig. 13) and this may, in some samples, make up more than 50 % of 559 the silcrete. Concretions generally show successive sequences of silica precipitation in which 560 there are alternations of thin laminae and thicker botryoidal layers of clear and brown opal. 561 The opal has been deposited uniformly around the voids, indicating a saturated groundwater 562 environment.

563 Crystalline varieties of silica occur also in these silcretes. They have formed by 564 recrystallisation of opal, either preferentially in some concretions as part of the sequence of deposition, or as in response to some form of alteration front (Fig. 13). The initial opal
deposits have often recrystallized to chalcedonite while the brown opal of the matrix has
transformed to microcrystalline quartz.

The opal deposits generally strengthen the silcrete and generate lustrous conchoidal fractures when knapped. However, when exposed at the landsurface in Australia, porcellanites can exhibit extensive crazing as a result of volume changes on dehydration of the opal phases.

572 4.4 Pedogenic silcretes

573 Pedogenic silcretes are highly variable in form and facies which results from the
574 diversity of the silicified parent materials (colluvium, clayey sand, clay silt, granite, bedrock
575 sandstone) and the position of the sample within the profile (columnar or nodular facies).
576 Adding to the variability is the mineralogical composition which records successive
577 recrystallization stages within the profile.

The most abundant and the most typical facies are tightly indurated by a dominantly microcrystalline quartz cement and have a conchoidal and lustrous fracture. Opal is mostly limited to less indurated facies at the base of the profiles and to specific micromorphological features within the profile that are always limited spatially and in volume.

The density and homogeneity of the microcrystalline quartz matrix can result in a material that is well-suited for knapping. The amoeboid microcrystals of quartz in net-like arrangement (Fig. 14) confers a structure and texture that mimics the characteristics of some flints. Silicified silty claystones that contain fine-grained detrital quartz and a substantial microcrystalline matrix can be knapped particularly well.

587 A principal criterion for recognising pedogenic silcretes is the omnipresence of 588 illuviation structures, particularly cutans at the base of voids and cappings over quartz grains 589 and granules. These are very clearly distinguished because they are outlined by opaque 590 microcrystalline titania (Fig. 14). As well, the common shard-like remnants of detrital quartz 591 grains point to significant dissolution and concomitant 'repacking' of the framework prior to 592 cementation, leading to a net volume loss in the profile as a whole.

5 593

Silcrete in landscapes

594 There are two aspects relating to silcretes in a landscape context. In the first case, 595 from the geological perspective, the occurrence and form of silcretes, particularly where 596 profile observations can be made, provides the basis for interpreting the environment of 597 formation of the silcrete. Macroscopic form and structure indicate the relative simplicity or 598 complexity of the processes of formation, point to single stage or overprinting characteristics, 599 and indicate conservation of volume and preservation of fabric and structure (groundwater 600 environments) or dissolution and loss of material with consequent and complex disruption in 601 the profile and, presumably, of the landsurface (pedogenic silcretes).

602 In the second case, within each category of silcrete, and even within a given category 603 in a particular location, the macroscopic and microscopic fabric and structure can vary 604 considerably. This is less so in the case of groundwater silcretes than for pedogenic silcretes. 605 Only particular forms of silcrete appear to have had value for the production of stone tools. 606 Mulvaney and Kamminga (1999), for example, point to the possibility that some forms of 607 silcrete could have been selected for tool production for aesthetic or symbolic reasons, 608 including their colour, which can be variable. Holdaway et al. (2008) and others (Doelman et al., 2001) make the distinction between 'outcrop silcrete' and the extensive lag gravels 609 ('gibber plains'), the latter (also called 'stony deserts') resulting from downwasting of the 610 611 ancient pedogenic silcrete-capped landsurfaces and associated groundwater silcrete-612 impregnated regolith over large areas of inland Australia (Fujioka et al., 2005). They point 613 out that artefacts were made from each source, but that outcrops were more difficult for 614 Aboriginal people to exploit for tools than gibber lags because of the greater supply of hand-615 sized cobbles in the latter.

616 Thus, if we want to explore the use of silcrete as a raw material for stone tools, for 617 example to map the routes of particular source materials and thus delimit territories and 618 exchanges between prehistoric groups of peoples, it is necessary to locate these materials in 619 the landscapes of the studied areas and the territories of the people exploiting them. In this 620 context, inventories of the locations of specific silicified formations, and their petrographic 621 characteristics, would be required for both local and regional areas. This research would 622 utilise an understanding of silcretes and silcrete facies in a landscape context because 623 prehistoric peoples had deeply-embedded connections with landscapes and their features.

624

# 5.1 Inland Australia

As is well known, pedogenic silcretes in Australia armour glacis (pediments) on the 625 626 edges of plateaux and around paleorelief (Milnes and Thiry, 1992; Milnes and Twidale, 1983; 627 Simon-Coinçon et al., 1996; Thiry et al., 1991; Twidale and Milnes, 1983a, b). At the edges 628 of scarps extending from the Arcoona Plateau in South Australia, for example, silicification is 629 restricted to thin 'skins', with a very high titania content, coating joint fractures and other surfaces on quartzite bedrock (Hutton et al., 1972; Fig. 15). On the proximal part of the Beda 630 631 pediment, just downslope from the scarp-foot zone of the plateau, silcrete is more extensive 632 and almost only formed of thick cappings (5 to 20 cm thick) on bouldery quartzite scree, 633 whereas in the distal areas of the pediment silcrete forms a regular horizon with columnar 634 structure within the pediment. Thicker silcretes occur in the transition zones between the 635 glacis and the lowland plains, where water discharge is still important but flow rates have 636 slowed.

The facies with thick cappings on scree boulders is symptomatic of the complexity that can face any archaeological search for the source of silcrete tools. These cappings are of very hard silcrete with a convex lustrous break and have been exploited for tools as shown by the litter of flakes and chips surrounding the outcrops. They may have been sought specifically because of certain characteristics. However, this particular facies is of very limited extent and locating it as a source of particular stone tools would only be possible after a very detailed inventory of a wide area.

In localities along the scarp of the Stuart Range, near Coober Pedy, in northern South
 Australia, a comprehensive analysis of the geomorphology of the region (Simon-Coinçon *et al.*, 1996) delineated the age relationships between the different weathering and silicification

features (Fig. 16). The pedogenic silcrete armours a wide paleopediment (tableland) dipping
from the Stuart Range near Lake Eyre to the Eucla Basin in the southwest, a distance of about
500 km. The weathered and bleached profiles are younger. Groundwater silcretes,
specifically quartzite and diverse porcellanite horizons, post-date the bleaching. Downcutting
and erosion ultimately generated the breakaway scarps now characteristic of the region.

652 The large variety of silicified materials in the Stuart Range landscapes are shown 653 schematically in more detail in Figure 16. Pedogenic silcretes may be locally buried beneath 654 younger deposits (clayey sediments and red-brown hardpans) but are omnipresent at or near 655 the surface of the widespread remnants of the Stuart Range tableland and available in 656 numerous outcrops and extensive lag gravels ('gibber plains', 'stony deserts'). The facies vary 657 according to the parent material affected by the silicification (coarse- or fine-grained; 658 abundant matrix material, or not, between the residual detrital grains). Illuviation structures 659 enriched in titania are symptomatic of these facies. To find any specific characteristics of a 660 particular facies in a particular location would seem to be difficult. On the other hand, 661 groundwater silcretes (quartzite, porcellanite and jasper of variegated colour) are available in 662 outcrops and lag gravels along the creeklines that leading down to Lake Eyre. They have more distinctive characters and a detailed inventory may provide tracers of specific materials. 663

664 About 700km north of Coober Pedy and the Stuart Range, in the Todd River Plain 665 near Alice Springs in central Australia, similar paleogeographic relationships occur. Here, in 666 the scarp foot zone of a formerly extensive early Cenozoic pediment developed over the Amadeus Basin, mesas and buttes capped by the massive and impressive columnar facies of a 667 668 pedogenic silcrete overlie a bleached and weathered regolith containing groundwater silcretes 669 (Milnes and Thiry, 1992; Milnes and Twidale, 1983). In distal parts of the pediment there 670 are plateau remnants armoured by thick groundwater silcretes and adjacent areas of silicified 671 limestones (Fig. 17).

Here, also, the silicified facies are extremely varied and the associated bleached
profiles are rich in white kaolinic clays and Fe oxides of variegated colours. Small mesas of
fine quartzitic sandstone have been intensively quarried for tool-making. The steep screeslopes around the hills and parts of the plateau surface are in places buried by thick deposits

of discarded flakes and chips. The clear evidence of significant (and possibly long-term)
exploitation provides the basis for characterising the source silcrete in detail and exploring the
distribution of stone tools made from it.

However, the vast region, maybe 1 million km<sup>2</sup>, stretching from central South Australia to the central parts of the Northern Territory, essentially marking the western part of the Lake Eyre Basin (see Alley, 1998; Fujioka *et al.*, 2005) has innumerable outcrops and derivative lag gravels of pedogenic and various groundwater silcretes, all of which could be exploited by Aboriginal people. Thus, there would be a significant challenge in progressively building an inventory of these materials sufficient to assist archaeological investigations in attributing artefacts to sources.

686 The nature and abundance of source material for tool-making has been noted from the 687 first archaeological works in inland Australia by Aiston (1928), who stated ".. it was so easy 688 to make a tool that directly one failed to work satisfactorily it was discarded and a fresh tool 689 made, a supply of stone material always being kept handy for this purpose...". The experience 690 of Australian archaeologists comes both from stone tools excavated from archaeological sites 691 and from ethnographic records of the experience of old Aboriginal people. Both types of 692 study concluded that Aboriginal people were highly opportunistic in their use of source stone 693 and took mostly varieties of local raw materials, including chert, chalcedony, jasper, silcrete 694 ('grey billy' in the vernacular), quartzite, basalt, silicified wood as well as other igneous and 695 metamorphic rocks (Cane, 1992; Flenniken and White, 1985). Nevertheless, small but 696 significant amounts of non-local materials have been found dispersed over long distances and 697 are thought to relate to migrations of prehistoric Aboriginal people during periods of drought 698 (Gould and Saggers, 1985; Holdaway and Fanning, 2014; Mulvaney and Kamminga, 1999).

699 5.2 Paris Basin

The Paris Basin is bounded by Jurassic marine limestone overlain by thick chalk deposits with alternating continental and marine Tertiary deposits in the centre. The most remarkable feature of the Tertiary sequence is the interbedding of sandstone formations with limestones and marls. The present-day morphology of the Paris Basin, namely superimposed limestone plateaux, results from major uplift during the Pliocene and Quaternary which
initiated downcutting and erosion of the Tertiary formations (Fig. 18). Silicified materials
occur in almost all formations.

707 Pedogenic silcretes extend from the peripheral basement in the Central Massif and 708 Britanny to the centre of the basin (Fig. 18). They developed along a paleosurface shaped by 709 clastic discharge during the lower Eocene (Blanc-Valleron and Thiry, 1997). Silicification 710 affects palaeoweathering profiles of clay-with-flint above Jurassic limestones and Cretaceous 711 chalks, old alluvium to form thick puddingstones, and sandy clays that become the 712 predominant silicified material at the border of the basin. The thickest silicified profiles (more 713 than 15m thick) are found in grabens and channels that formed the lowlands (Thiry and 714 Simon-Coinçon, 1996).

715 Groundwater silcretes developed in almost every Tertiary formation in the centre of 716 the basin. Tightly cemented quartzite lenses occur in all sandy formations and silicified zones 717 are present in every lacustrine and marine limestone formation (Thiry, 1999). Even the 718 weathered materials topping the limestone plateaux contain large silicified features. 719 Groundwater silcretes also developed in sandy and conglomeratic regolith materials topping 720 the Mesozoic sequences bordering the basin. Recent groundwater cementation has been 721 superimposed on former pedogenic silcrete profiles on the outer borders of the basin. It is 722 likely that 5 to 10% of the outcropping formations in the basin are silicified, and this 723 represents a considerable mass of easily accessible silcrete for potential exploitation.

Added to this inventory as well are Jurassic cherts and Cretaceous flints that occur all over the Paris Basin. They are *in situ* in the Mesozoic formations or occur in extensive areas covered by paleoweathering profiles of clay-with-flint and reworked into thick alluvial deposits within valleys that dissect the plateaux. These Jurassic cherts and Cretaceous flints generally account for the main component of lithic source material. Their identification was essentially based on micropaleontological determinations and was consequently limited in terms of pointing to source areas (Masson, 1981; Valensi, 1953).

Due to the predominance of cherts and flints, less interest has been applied to the
recognition of various silcrete facies. Nevertheless, some investigations, in particular in the

733 Paris Basin, identified several kinds of silcretes used for making tools. Fine-grained 734 quartzitic silcretes from Fontainebleau have been recognized in flakes (Robin, 1974), used for 735 polished axes (Bostyn et al., 2012), and even to made into thin leaf-shaped Solutrean points 736 (Sacchi et al., 1996). These silcretes correspond to the uppermost facies, directly below the 737 limestone cover, and are of glassy appearance due to the presence of microquartz together 738 with quartz overgrowths forming the cement. Bartonian silicified lacustrine limestone has 739 also been recognized (by means of its fossil content) in polished axes that are thought to have 740 been dispersed up to 250 km (Bostyn *et al.*, 2012). Elsewhere, similar Bartonian silicified 741 lacustrine limestone has been found together with artefacts made from meulière (weathered 742 silicified limestone), silicified Lutetian marine limestone and quartzitic silcretes (Augereau, 743 2008; Lanchon et al., 2008; Mauger, 1985; Surmely, 2009).

744 Recent studies in and around the margins of the Massif Central, taking into account 745 the petrography of cherts, flints and silcretes, including changes in response to their colluvial 746 and alluvial reworking, made it possible to refine source areas (Fernandes, 2012; Surmely et 747 al., 2008). It has to be stressed that it is in basement areas (Massif Central and Bretagne) 748 where the siliceous materials are scarce that the sourcing studies were the more skilful and the 749 most successful. Detailed studies of flint and chert cortices from the Massif Central made it 750 possible to consider natural dispersion of the primary materials and consequently to determine 751 more precisely the source areas (Fernandes, 2012). Silcrete deposits exist also in these 752 basement areas, although scattered and variable in facies. Detailed studies have been able to 753 track the dissemination of these materials within the massif as well as those imported from the 754 peripheral basins (Aubry, 1991; Célérier, 1990; Dabard et al., 2012; Delvigne, 2012; Surmely 755 et al., 2008; Wragg-Sykes, 2014).

#### 756 6 Identifying provenance

There are certainly specific macroscopic criteria that could be used to help identify sources of silcrete tools, but most of the time these investigations are ambiguous. Colour, grain size and distribution, and the character and aspects of the break, are generally not specific. The morphologies of silcretes are often typical at the outcrop scale, but not at the artefact scale.

762 Micromorphological and petrographical features of silcretes are more reliable criteria 763 of origin and potential provenance. However, identification of provenance from these 764 features requires a comprehensive inventory of actual local and regional source materials, a 765 sound geological knowledge of these occurrences, and an accompanying database of 766 micromorphological and petrographical observations of representative samples. This is 767 conceivable on a local scale but is much more difficult to manage on the regional scale of 768 societal exchanges. Geochemical criteria, particularly the concentrations of indicator trace 769 elements, may be helpful. Dutkiewicz et al. (2015) have analysed major and (particularly) 770 trace elements in both opal and host rocks in Australia in order to identify the possible origin 771 of the silica. This approach could potentially be extended to silcretes although, in pedogenic 772 silcretes, the complex of silica forms and associations with microcrystalline anatase and other 773 phases, and the multiplicity of silcrete facies, even on the scale of an outcrop, would pose a 774 significant challenge. Moreover, Holdaway and Fanning (2014) and others have pointed out 775 the complexities presented by the suite of stone artefacts in archaeological sites wherein there 776 are local and foreign sourced materials representing an interplay of unknowable activities.

Finally, questions of the extent of post-discard alteration and degradation of
archaeological materials has to be addressed. Exoscopic studies of the surfaces of stone tools,
which are often altered in various ways by patina evolution, can provide few critical data
(Fernandes, 2012; Thiry *et al.*, 2014).

The difficulty for archaeological studies is to precisely characterize the many varieties of silica-rich source rocks for stone tools. In the past, only stratigraphic characters and fossils have been used to determine the provenance of tool materials (Mauger, 1984). This was of limited use because the stratigraphic levels are relatively widespread. More detailed, multicriteria identification is necessary and essential for understanding trade and acquisition strategies involving stone tools.

Progress is possible by taking into account precise micromorphological characteristics that will potentially point to source sites. However, this approach requires an insightful inventory and petrographic data led by archaeological studies (Fernandes, 2012). Inventories of this type are yet not available. Even existing geologic and geomorphological maps do not make reference to the specific characters of silcretes, silcrete profiles, or micromorphological
features of particular facies, that could be key to these types of investigations. The
foundations of such inventories exist, but a new generation of linked archaeological and
geological research would be a next step.

795 **7 Summary** 

796 Several forms of silcrete are recognised in remnants of paleolandscapes in Europe, 797 Africa and Australia dating from the Mesozoic to the present day. Their chemical, 798 mineralogical and physical (microstructural) composition accounts for their persistence and 799 for the major influence of some silcretes in defining the unique morphology of landforms. All 800 silcretes, of which there are 4 - 6 main types, formed as a result of the introduction and 801 precipitation of silica, for example via groundwaters, or the mobilisation and precipitation of 802 silica from pre-existing minerals via alteration and the leaching of other elements, for 803 example in acidic environments. Each is characterised by a variety of silica polymorphs, 804 including quartz, various forms of microcrystalline and cryptocrystalline quartz, and various 805 forms of opal (opal-CT, opal-A), in particular micromorphological arrangements. The main 806 categories of silcrete originated in pedogenic or groundwater environments.

807 Other than their mode of occurrence and morphological appearance, it is the 808 microstructural composition of silcretes, dictated by mineralogical and micromorphological 809 features that derive from their origin, that provides the basis for understanding their physical 810 and mechanical properties (which relates to the attractiveness of some forms to prehistoric 811 peoples for the making of effective tools). However, it is well recognised that there may have 812 been other practical and even aesthetic or symbolic dimensions to such choices. 813 Investigations of the provenance of stone tools, other than in a local area, will be challenged 814 by a need for regional inventories of silcrete facies.

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- 816

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#### 1116 Figure captions

1117

1118 Figure 1 - Rate of dissolution and asymptotic solubility of the different silica varieties at 25°C

(after Siffert 1967). Opal and chalcedony show variable solubility according to theirrespective crystallinity.

Figure 2 - Quartz and chalcedony solubility versus temperature (computed from geochemist's workbench program; Bethke, 2002). Quartz solubility decreases to more than half of its initial value between 12.5 and 0°C.

Figure 3 – Solubility of amorphous silica in aqueous salt solutions at 25°C (after Marshall,
1125 1980)

Figure 4 – Alumina and silica solubility versus pH (computed from PHREEQC program, U.S.
Geological Survey; Parkhurst and Appelo, 1999). Silica is insoluble at acidic and neutral
pH that appear favourable for development of silcretes, whereas alumina stability is
restricted to neutral pH.

1130 Figure 5 – Eh/pH diagram for Fe-O-H system showing the reaction path leading to ferrolysis

1131 after having introduced some oxygen into the  $Fe^{++}$  rich solution. Diagram at 25°C in

equilibrium with atmospheric  $O_2$  and  $CO_2$ , and  $[SO_4--]=10^{-7}$  (after Garrels and Christ, 1133 1965).

Figure 6 –Schematic geological section through the Beauce Plateau. Quartzitic silcrete pans
are limited to the outcrop zone of the Fontainebleau Sand. The size of the silcrete lenses
(black wedges) is exaggerated for illustration.

Figure 7 – Schematic model of successive cycles of groundwater-related quartzitic silcrete
formation in the Fontainebleau Sand. The thickness and the slope of quartzite pans are
exaggerated for illustration.

1140 Figure 8 - Schematic diagram of the Calcaire de Champigny Formation (upper Eocene)

1141 showing the distribution and shapes of silicified zones (Plateau of Brie, France). Note

similarity between the distribution and shapes of the silicified zones and dissolution

features of the limestone.

1144 Figure 9 - Larkins Folly section exposed in bulldozer costeans in Coober Pedy opal field. The 1145 bleached Cretaceous formations contain termite burrows and alunite nodules. There are 1146 two superposed levels of groundwater silicification. In addition, at the top is a 1147 pedogenic silcrete disrupted by vertical pipe-like structures, and an overlying laminar-1148 structured red-brown hardpan. 1149 Figure 10 – Sketch of the macromorphological organisation of a pedogenetic quartzose 1150 silcrete and distribution of secondary silica and dissolution structures. 1151 Figure 11 – Thin sections of groundwater silcretes (transmitted light, crossed polars). (A) 1152 Fontainebleau quartzitic silcrete in which original detrital quartz grains are overgrown 1153 by secondary silica, producing triple-point junctions and thus a compact mesh that 1154 results in a rough fracture. (B) Stuart Creek quartzitic silcrete with chalcedonite deposits 1155 in pore spaces produces a lustrous break on knapping. 1156 Figure 12 – Thin sections of silicified limestone (Paris Basin): two views (A) and (B): 1157 transmitted light, crossed polars. The larger crystals are euhedral quartz that crystallized 1158 in voids (v) by precipitation from water, the speckled areas are microcrystalline quartz 1159  $(\mu Q)$  replacing primary carbonate and the black areas are remnant carbonate (Ca). The 1160 images show the nature of the silicified "front" (dotted line). 1161 Figure 13 – Thin sections of porcellanites (transmitted plane polarised light). (A) Stuart 1162 Creek, South Australia. The dark granular matrix (Op1) is silicified silty claystone. 1163 Voids (v) and channels contain successive silica deposits (Op2), firstly brown opal then 1164 chalcedonite. The latter also impregnates the matrix and results in a compact material 1165 with a lustrous fracture. (B) Sancerre, southern Paris Basin. The dark opal matrix (Op1) 1166 has recrystallized into clear homogeneous microquartz ( $\mu Q$ ); the opal deposits in the voids (Op2) remain unaffected. This type of porcellanite provides sharp flakes, like 1167 1168 flints, on knapping. 1169 Fig 14 – Thin sections of pedogenic silcretes, Paris Basin. (A) matrix of amoeboid quartz 1170 microcrystals in net-like arrangement generates a hard and lustrous fracture on knapping 1171 (transmitted light, crossed polars). (B) and (C) nodular facies with illuviation cutans (il) 1172 recrystallized into microquartz plus titania microcrystals; nodules are rimmed by a

1173	cortex enriched in titania (arrow). Channels remain empty (v) or are cemented with sub-
1174	euhedral quartz crystals (Q). (Transmitted plane polarised light).
1175	Figure 15 - Schematic view of the Beda pediment showing the distribution of pedogenic
1176	silcrete facies. Facies change completely at a 100 m scale (after Milnes and Thiry,
1177	1992).
1178	Figure 16 - Schematic morphostratigraphic sketch of landscapes from the Davenport Range,
1179	across the Stuart Range, to the Eucla Basin, South Australia, showing silicification in
1180	relation to different weathering features and their distribution in relation to landsurfaces
1181	and geology (after Simon-Coinçon et al., 1996).
1182	Figure 17 –Schematic section across the Todd River Plain, southeast of Alice Springs, central
1183	Australia, showing the location of different silcrete facies (after Milnes and Thiry,
1184	1992).
1185	Fig. 18 – Schematic geomorphological section across the Paris Basin, showing the
1186	distribution and relationships of silcrete facies.
1187	





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